Neutralization control device

TK SOMAK

The pH of the alkaline treatment tank is monitored and diluted acid is automatically injected.

Automatic control to set pH value!





[CHARACTER]

Neutralization is performed with two types of acids, organic acid and inorganic acid.

Utilizes the difference in reaction rate

between organic and inorganic acids.

Enables safe and inexpensive neutralization!

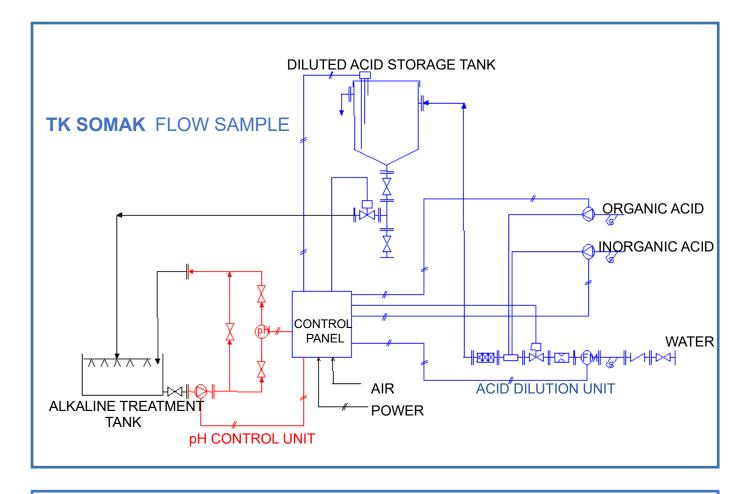


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[TK SOMAK Organic acid / inorganic acid combined neutralization control device]

Organic acid \Rightarrow Safe, but expensive. Also, the BOD value is high.

Inorganic acid \Rightarrow Cost is low, but there is a risk of fabric embrittlement.

TK SOMAK solves these problems by taking advantage of the difference in reaction rate between organic and inorganic acids.

Most of the neutralization is done with a fast-reacting inorganic acid, and the rest is completed with a slow-reacting organic acid.

Even if there is acid residue, it will be an organic acid without the risk of dough embrittlement.

[Example] Consider the case where caustic soda (NaOH) is neutralized

with sulfuric acid (H2SO4) and formic acid (HCOOH).

- 1. Sulfuric acid and formic acid react separately with NaOH at the same time.
- 2. Sulfuric acid reacts with formic acid (HCOONa sodium formate) that has reacted with NaOH and returns to formic acid.
- 3. Formic acid reacts with NaOH.
- 4. Same as ②, repeat below. In other words, formic acid always remains after the reaction of sulfuric acid.

TK SOMAK enables safe and inexpensive acid neutralization.